

A Cold-Start Free Fuel Cell Fed with a Rechargeable “Gasoline”

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Extended Abstract

In the last decades, driven by the depletion of conventional fossil fuels, the mission for finding alternative systems for sustainable power generation becomes more and more urgent [1]. Among the diverse kinds of novel power generation systems, the direct liquid fuel cells, are believed to be one of the most important candidates [2]. However, their poor performance at low temperatures, particularly in the sub-zero environment, greatly restricted their world-wide applications [3]. Recently, a novel rechargeable “gasoline”, named the electrically rechargeable liquid fuel (e-fuel), has been proposed and proved to be an effective approach for storing intermittent renewable power [4]. In comparison to the conventional liquid fuels, this rechargeable “gasoline” has many advantages including: i) wide material selectivity; ii) rechargeability; and iii) fast reaction kinetics even on catalyst-free electrode. Furthermore, it also possesses a freezing point of as low as $\sim -30^{\circ}\text{C}$. [5]

In this work, fed with this rechargeable “gasoline”, we develop, characterize and demonstrate a passive liquid fuel cell capable of stably generating electricity in a sub-zero environment. By applying a cold-start-free strategy, where the cell does not need any auxiliary/internal equipment for rapid warm up at the beginning of its operation, the cell temperature is able to be maintained at sub-zero while being able to achieve a stable and high performance. It is found that, utilizing the rechargeable “gasoline”, the cell is able to achieve a peak power density of 110.34 mW cm^{-2} and a maximum current density of 244.72 mA cm^{-2} even at -20°C , demonstrating it as a highly attractive candidate for applications under a wide range of operating temperature. To the best of our knowledge, neither traditional direct liquid fuel cells nor hydrogen-oxygen fuel cells have ever been observed to operate successfully under the aforementioned operating conditions, while a great improvement is also achieved even in comparison to our previous work [5]. This important breakthrough is thus believed to provide an important direction for the advancement of fuel cell technology for operation under sub-zero temperatures, especially for supplying future FCEVs under all-climate conditions.

References

- [1] C. Duan, R. Kee, H. Zhu, N. Sullivan, L. Zhu, L. Bian, D. Jennings and R. O Hayre. “Highly efficient reversible protonic ceramic electrochemical cells for power generation and fuel production”. *Nat. Energy*, 4, 230-240, 2019.
- [2] B. Ong, S. Kamarudin and S. Basri, “Direct liquid fuel cells: A review”. *Int. J. Hydrogen Energy*, 42, 10142-10157, 2017.
- [3] C. Bianchini and P. K. Shen. “Palladium-based electrocatalysts for alcohol oxidation in half cells and in direct alcohol fuel cells”. *Chem. Rev.*, 109, 4183-4206, 2009.
- [4] H. Jiang, L. Wei, X. Fan, J. Xu, W. Shyy, and T.S. Zhao. “A novel energy storage system incorporating electrically rechargeable liquid fuels as the storage medium”. *Sci. Bull.*, 64, 270-280, 2019.
- [5] X. Shi, X. Huo, O.C. Esan, Y. Ma, L. An, and T.S. Zhao. “A liquid e-fuel cell operating at -20°C ”. *J. Power Sources*, 506, 230198, 2021.