

Impact of BiBO₃, NaF and BiF₃ Doping on the Thermoelectric Properties of Bi₂Ca₂Co₂O_y Ceramics

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Extended Abstract

Due to the increasing global energy demand and climate change issues, the importance of environmentally friendly renewable energy technologies is growing. Many studies in this field focus on thermoelectric materials that can directly convert waste heat into electric power. Doping via sol-gel chemistry, leading to exceptionally high chemical homogeneity, is a promising approach for enhancing the heat-to-electricity conversion efficiency of thermoelectric materials. In this work, for the first time, we studied the impact of (i) partial substitution of bismuth oxide (Bi₂O₃) by bismuth borate (BiBO₃), (ii) dual-substitution of Bi₂O₃ by BiBO₃ and sodium fluoride (NaF), and (iii) dual-substitution of Bi₂O₃ by BiBO₃ and bismuth (III) fluoride (BiF₃) on the thermoelectric characteristics of Bi₂Ca₂Co₂O_y layered cobaltite. Samples with a nominal composition of Bi₂Ca₂Co₂O_y (reference), Bi_{1.9925}Ca₂Co₂(BiBO₃)_{0.0075}O_y, Bi_{1.8925}Ca₂Co₂(NaF)_{0.10}(BiBO₃)_{0.0075}O_y, and Bi_{1.8925}Ca₂Co₂(BiF₃)_{0.10}(BiBO₃)_{0.0075}O_y, were prepared using the sol-gel method with citric acid and ethylene glycol as chelating agents. The phase composition of prepared materials was examined using X-ray diffraction analysis. Grain size distribution, surface morphology, and elemental compositions were studied by a scanning electron microscope (SEM) coupled with an energy-dispersive X-ray spectrometer (EDS). A homemade setup was used to determine the temperature dependence of the Seebeck coefficient and electrical resistivity in the temperature range of 293 K to 973 K. Thermal conductivity was measured from 293 K to 573 K. Finally, values of power factor ($PF = S^2/\rho$) and figure of merit ($ZT = S^2T/\rho k$) were calculated with S , ρ , T , and k representing Seebeck coefficient, electrical resistivity, absolute temperature, and thermal conductivity, respectively. The XRD analysis confirms that all the samples consist of a nearly pure Bi₂Ca₂Co₂O_y phase. SEM/EDS analysis reveals that prepared materials consist of plate-like grains with a sizes of around 6 μm , which constitute the Bi₂Ca₂Co₂O_y phase. Additionally, there are smaller grains with dimensions of 1–2 μm , mainly containing a cobalt compound. When compared to the reference and BiBO₃-doped samples, dual NaF/BiBO₃ and BiF₃/BiBO₃ doping significantly reduces the concentration of smaller grains. Dual NaF/BiBO₃ doping leads to a sharp decrease in ρ due to the partial substitution of NaF for Bi₂O₃, which increases the concentration of charge carriers (holes). At the same time, partial substitution of BiF₃ for Bi₂O₃ led to a decrease in hole concentration and, hence, an increase in ρ . Seebeck coefficients were positive for all the samples, indicating p -type conductivity. The value of S for the NaF/BiBO₃ co-doped composition at 973 K is 4–6 % lower than that of the reference, BiBO₃-doped, and BiF₃/BiBO₃ co-doped samples. Although the NaF/BiBO₃ co-doped sample possesses a slightly lower Seebeck coefficient and higher thermal conductivity than the ones studied, the reduced resistivity has a greater impact on the thermoelectric performance, leading to an improved power factor and figure of merit. The maximum PF and ZT values achieved in the NaF/BiBO₃ co-doped composition are 18% and 13% higher, respectively, than the reference Bi₂Ca₂Co₂O_y.

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