

# Numerical Analysis of Gas Diffusion Characteristics during Thermal Runaway in Lithium-Ion Battery Module

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## Extended Abstract

Lithium-ion batteries are widely used as secondary batteries due to their high energy density and low self-discharge. However, the lithium-ion batteries have the risk factor, such as explosions and toxic gas emissions caused by thermal runaway. Thermal runaway is a phenomenon where the temperature of the battery rapidly increases due to external factors, which leads to chemical reactions inside the battery, and the generation and ejection of various gases such as H<sub>2</sub>, CO, CO<sub>2</sub>, and VOCs to the outside.[1,2] Analyzing the vented gas flow from the battery is essential for designing fire safety devices, such as early detection systems for battery fires. The CO<sub>2</sub> gas is known most advantageous for early detection of thermal runaway [3]. However, most previous studies on gas flow analysis have been performed in single cells, while lithium-ion batteries are typically used as battery modules.[4,5] Therefore, this study was used to analyse the gas diffusion inside the module where thermal runaway of the battery occurred. This study performed numerical analysis using the commercial program Ansys fluent 19.1. The standard k-ε model is used for gas diffusion in the battery module and the species transport model to simulate the gas generated from the battery. The analysis was unsteady, with a time step of 0.01 s, and set for an analysis time of 100 s. The results show that at the beginning of gas ejection, the gas diffuses rapidly to adjacent cells where thermal runaway occurs. The gas velocity distribution before 7 seconds after thermal runaway describes the gradual diffusion of the gas into the module. However, the amount of gas initially ejected is small, and the CO<sub>2</sub> mass fraction distribution before 10 seconds is low. After 8 seconds, there is a region where the gas diffusion rate suddenly increases. This is because of the flow that hits the module wall due to the vortex, and the flow moves into the center of the module due to the vortex. After 10 seconds, the released CO<sub>2</sub> gas gradually accumulates at the end of the module and diffuses throughout the interior.

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