Biobased Latexes from Soy and Vanillin Derived Vinyl Monomers

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Extended Abstract

Recently, we reported on synthesis and characterization of new biobased acrylic monomers derived from variety of plant/vegetable oils (soybean, olive, linseed, sunflower, corn, canola etc.) and their application in free radical polymerization including emulsion/miniemulsion latex synthesis, with a goal to replace petroleum-based constituents and reduce VOC content.^{1,2} Copolymerization study on plant oil-based monomers (POBMs) with a variety of petroleum-based commodity comonomers demonstrated that incorporation of POBM fragments into macromolecular structure provides internal plasticization and hydrophobization, improves toughness of crossilinked latex polymers, enhances adhesion properties and promotes biodegradability of resulted biobased latexes. However, excessive polymers softening was observed when POBM content in the resulted copolymers exceeded 60 wt.%, thus making latexes "too soft" and not applicable for film preparation. To be able to further increase biobased content in latexes without worsening the material properties and performance, biobased aromatic comonomers have to be considered for copolymerization with POBM. We hypothesize that combination of a benzene ring (present in the structure of aromatic monomer) and long fatty acid fragments (present in POBMs) provides a lot of opportunities in terms of tuning latex copolymers properties and performance. To this end, vinyl monomer derived from aromatic cardanol (CBM) and monomer from high oleic soybean oil (HOSBM) were successfully copolymerized in miniemulsion to yield biobased latexes in our recent study.³ As a result, both biobased fragments contributed to the mechanical properties of the synthesized latexes. The incorporation of aromatic fragments of cardanol enhances Young's modulus of the films, whereas the fatty POBM constituents made the material more flexible. Already 10 wt. % of CBM fragments provided noticeable strength to the soft latex polymeric material from entirely HOSBM.³

Furthermore, we elaborated on hypothesis of combining aromatic and aliphatic biobased ingredients in emulsion/miniemulsion copolymerization process. For this purpose, new vanillin-derived compound, 3-allyl-5-vinylveratrole (AVV)⁴ was chosen as a comonomer for copolymerization with HOSBM. Having chemical structure similar to petroleum-based commodity vinyl monomer styrene, the AVV can be considered as styrene's biobased alternative when incorporated in a variety of polymer materials. Incorporation of AVV fragments into latex copolymers could balance mechanical properties of the resulted macromolecules by adding the materials strength and enhancing toughness. Additional interesting aspect can be added by presence of allyl bond in the AVV chemical structure which sets task to investigate its polymerizability in free radical mechanism, specifically, whether the allyl group remains retained during the reaction and can be used for post-polymerization cross-linking.

The obtained results show that AVV successfully undergoes free radical homopolymerization ($M_n=2$ 100 000) and chain copolymerization. The DSC of polyAVV showed that the synthesized biobased material possesses a T_g in the region of 65 – 70°C. Finally, feasibility of AVV and HOSBM to interact in chain copolymerization was demonstrated. Both fragments contribute to the properties of resulted latex copolymers. If compared to HOSBM copolymerization with CBM, significantly higher molecular weight copolymers can be obtained. This can be explained by the fact that pronounced for more unsaturated CBM allylic termination, thus chain propagation coexisting with effective chain transfer, did not take place when HOSBM is copolymerized with AVV. As determined by ¹H NMR spectroscopy, AVV's allylic group remains unreacted under polymerization reaction conditions used in this study.

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