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## Removal of Arsenic from Aqueous Solutions using Natural Biochar-Based Sorbent from Fibers of Date Palm (*Phoenix dactylifera*)

Aisha A. Al-Busaidi<sup>1</sup>, Mushtaque Ahmed<sup>2</sup>, Hassan. A Al-Reasi<sup>3</sup> Yasmine Souissi<sup>4</sup>

<sup>1</sup>University of Technology and Applied Sciences
Al-Khuwair, Muscat, Oman
Aisha.albusaidi@utas.edu.om; ahmedm@squ.edu.om

<sup>2</sup>Sultan Qaboos University, Muscat, Oman
HReasi@su.edu.om

<sup>3</sup>Sohar University, Sohar, Oman
Yasmine. Souissi@gutech.edu.om

<sup>4</sup>German University of Technology in Oman, Muscat, Oman

## **Extended Abstract**

Biochar (BC) is a durable, low-cost sorbent for aqueous metal removal; manganese-oxide functionalization (via KMnO<sub>4</sub>) enhances arsenic uptake by introducing redox-active MnOx domains and additional oxygenated surface sites that favor As (III) oxidation/adsorption [1], [3],[8]. This study reports an enhanced biochar (EBC) produced from KMnO<sub>4</sub>-treated date-palm fiber by slow pyrolysis, benchmarked against pristine BC. Objectives were to (i) increase specific surface area and (ii) enrich oxygen-containing functionalities/MnOx domains to strengthen As (III) removal. The scope covered synthesis, physicochemical characterization, batch adsorption under environmentally relevant conditions, isotherm modeling, and regenerability.

Materials were characterized by SEM–EDS, BET N<sub>2</sub> physisorption, FTIR, and ICP–OES to quantify morphology, texture, functionality, and composition [2]. Relative to BC, EBC exhibited larger surface area and higher densities of oxygenated groups with detectable Mn-bearing deposits, consistent with oxidative modification and MnOx loading [1], [3], [4]. Batch tests evaluated removal efficiency (E, %) and adsorption capacity (q<sub>e</sub>, mg·g<sup>-1</sup>) for As (III) at pH 3, 7, 9; initial concentrations 5, 50, 100 mg·L<sup>-1</sup>; and sorbent doses 0.05, 0.20, 0.40 g per 0.025 L. EBC consistently outperformed BC, reaching a maximum E of ~70% versus ~40% for BC. Observed pH, concentration, and dose effects follow trends reported for Mn-modified biochars and date-palm–derived biochars used in metal/arsenic remediation [1], [4], [9], [12].

Equilibrium data favored the Langmuir model over Freundlich (representative R<sup>2</sup> = 0.97), indicating monolayer adsorption on a finite set of relatively uniform sites—consistent with the creation of new high-affinity MnOx/oxygenated centers during KMnO<sub>4</sub> treatment [5]. Regeneration using aqueous NaOH restored EBC performance with minimal capacity loss across cycles, aligning with evidence that alkaline desorption effectively reactivates metal-loaded biochar adsorbents [6], [13]. Overall, KMnO<sub>4</sub>-assisted modification of date-palm–fiber biochar improved As (III) removal by coupling textural enhancement with targeted surface chemistry, offering a practical route to valorize regional agricultural residues into regenerable sorbents for wastewater treatment [1]–[4],[12].

**Keywords:** biochar; date-palm fiber; KMnO<sub>4</sub> modification; MnOx; arsenite; adsorption capacity; removal efficiency; Langmuir isotherm; regeneration.

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