Proceedings of the 7th International Conference of Recent Trends in Environmental Science and Engineering (RTESE'23) Ottawa, Canada – June 04 - 06, 2023 Paper No. 172 DOI: 10.11159/rtese23.172

Removal of Pharmaceuticals from High-Strength Wastewater by Adsorption on Commercial Granular Activated Carbon: Study of the Operating Conditions

Mina Asheghmoalla, Mehrab Mehrvar

Department of Chemical Engineering, Toronto Metropolitan University 350 Victoria St, Toronto, Ontario, Canada M5B 2K3 mina.asheghmoalla@torontomu.ca; mmehrvar@torontomu.ca

The increased need for pharmaceuticals and personal care products has resulted in a rise in the volume of wastewater produced by drug manufacturing facilities. The manufacturing methods employed in producing pharmaceuticals result in the generation of significant amounts of heavily contaminated wastewater. These pollutants, if released into nature, can pose significant risks to both the environment and human health [1]. Sulfa drugs, aniline-derived compounds and solvents are among the organic matters that are used or generated during drug synthesis and are subsequently introduced into the wastewater pipeline during the vessel wash and system shutdown operation [2,3].

Many of pharmaceuticals from this industry are considered emerging contaminants, and their concentrations in surface water are limited by national, federal, or provincial regulations. The toxic effects and non-biodegradable nature of these chemicals highlight the urgent need for proper wastewater treatment techniques to mitigate their harmful impact [4,5]. Conventional wastewater treatment methods, such as biological and chemical treatments, are not sufficient for removing pharmaceuticals, as they can be costly, time-consuming, and inefficient in removing these micropollutants [6,7].

To address the removal of emerging contaminants from wastewater prior to discharge into the environment, advanced treatment technologies such as advanced oxidation processes, activated carbon adsorption, and membrane filtration need to be implemented [8]. Among advanced treatment methods, adsorption by activated carbon is beneficial and preferred over other techniques due to the operation simplicity, no by-product production, low process cost, and high efficiency in the removal of micropollutants from water [9,10]. Nevertheless, the application of this process to the wastewater domain is limited due to the high concentration of pollutants in industrial wastewater. Therefore, the use of activated carbon adsorption for the removal of pharmaceuticals from high-strength industrial wastewater requires further exploration to optimize its effectiveness [11,12].

In this study, it is desired to investigate the effect of pH and activated carbon dosage on the removal of six selected pharmaceuticals from a medium-high strength synthetic wastewater, which in terms of properties, resembles the actual effluent of drug manufacturing facilities. To meet the objectives of this study, a specific amount of commercial granular activated carbon was mixed with synthetic wastewater, and the total organic carbon (TOC), which represents the total amount of pharmaceuticals in the wastewater, was measured. The batch adsorption studies showed that the removal increases by increasing the dosage of activated carbon and that higher removal can be achieved at acidic pH. The results of this study will create a baseline to optimize the adsorption process by adjusting the pH and mass of activated carbon required for a certain degree of removal.

Keywords: Pharmaceuticals removal; activated carbon adsorption; wastewater treatment; operating conditions; aniline removal; high-strength wastewater

References

- [1] S. Ghafoori, A. Mowla, R. Jahani, M. Mehrvar, and P. W. H. Chan, "Sonophotolytic degradation of synthetic pharmaceutical wastewater: Statistical experimental design and modeling," *Journal of Environmental Management*, vol. 150, pp. 128–137, Mar. 2015, doi: 10.1016/j.jenvman.2014.11.011.
- [2] S. K. Al-Dawery, "Adsorption of methanol from methanol-water mixture by activated carbon and its regeneration using photo-oxidation process," *Desalination and Water Treatment*, vol. 57, no. 7, pp. 3065–3073, Feb. 2016, doi: 10.1080/19443994.2014.980331.

- [3] H. Koyuncu and A. R. Kul, "Removal of aniline from aqueous solution by activated kaolinite: Kinetic, equilibrium and thermodynamic studies," *Colloids and Surfaces A: Physicochemical and Engineering Aspects*, vol. 569, pp. 59–66, May 2019, doi: 10.1016/j.colsurfa.2019.02.057.
- [4] R. Jahani, R. Dhib, and M. Mehrvar, "Photochemical degradation of aqueous artificial sweeteners by UV / H₂ O₂ and their biodegradability studies," *Journal of Chemical Technology & Biotechnology*, vol. 95, no. 9, pp. 2509–2521, Sep. 2020, doi: 10.1002/jctb.6432.
- [5] Y. Sun, Y. Wang, Z. Peng, and Y. Liu, "Treatment of high salinity sulfanilic acid wastewater by bipolar membrane electrodialysis," *Separation and Purification Technology*, vol. 281, p. 119842, Jan. 2022, doi: 10.1016/j.seppur.2021.119842.
- [6] K. Arola, H. Hatakka, M. Mänttäri, and M. Kallioinen, "Novel process concept alternatives for improved removal of micropollutants in wastewater treatment," *Separation and Purification Technology*, vol. 186, pp. 333–341, Oct. 2017, doi: 10.1016/j.seppur.2017.06.019.
- [7] A. Mowla, M. Mehrvar, and R. Dhib, "Combination of sonophotolysis and aerobic activated sludge processes for treatment of synthetic pharmaceutical wastewater," *Chemical Engineering Journal*, vol. 255, pp. 411–423, Nov. 2014, doi: 10.1016/j.cej.2014.06.064.
- [8] X.-T. Bui, T. P. T. Vo, H. H. Ngo, W. Guo, and T. D. Nguyen, "Multicriteria assessment of advanced treatment technologies for micropollutants removal at large-scale applications," *Science of the Total Environment*, vol. 563–564, pp. 1050–1067, Sep. 2016, doi: 10.1016/j.scitotenv.2016.04.191.
- [9] J. R. De Andrade, M. J. Oliveira, M. G. C. Da Silva, and M. G. A. Vieira, "Adsorption of Pharmaceuticals from Water and Wastewater Using Nonconventional Low-Cost Materials: A Review," *Industrial & Engineering Chemistry Research*, vol. 57, no. 9, pp. 3103–3127, Feb. 2018, doi: 10.1021/acs.iecr.7b05137.
- [10] M. A. Zamiri and C. H. Niu, "Development and characterization of novel activated carbons based on reed canary grass," *Industrial Crops and Products*, vol. 187, p. 115316, Nov. 2022, doi: 10.1016/j.indcrop.2022.115316.
- [11] F. Mansour, M. Al-Hindi, R. Yahfoufi, G. M. Ayoub, and M. N. Ahmad, "The use of activated carbon for the removal of pharmaceuticals from aqueous solutions: a review," *Reviews in Environmental Science and Bio/Technology*, vol. 17, no. 1, pp. 109–145, Mar. 2018, doi: 10.1007/s11157-017-9456-8.
- [12] M. Eljaiek-Urzola, L. Guardiola-Meza, S. Ghafoori, and M. Mehrvar, "Treatment of mature landfill leachate using hybrid processes of hydrogen peroxide and adsorption in an activated carbon fixed bed column," *Journal of Environmental Science and Health, Part A*, vol. 53, no. 3, pp. 238–243, Feb. 2018, doi: 10.1080/10934529.2017.1394709.